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Key indicators

Single-crystal X-ray study
 $T = 93\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$
 R factor = 0.097
 wR factor = 0.309
Data-to-parameter ratio = 12.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.*N,N'*-Bis(4-pyridyl)naphthalene-3,4:7,8-di-
carboximide dimethylformamide disolvateIn the title compound, $\text{C}_{24}\text{H}_{12}\text{N}_4\text{O}_4 \cdot 2\text{C}_3\text{H}_7\text{NO}$, the naphthalene imide, which is an organic pigment utilized for H_2 gas sensors, has inversion symmetry. The pyridyl rings are twisted, in the same direction, out of the plane of the naphthalene imide skeleton by $69.5(2)^\circ$.

Comment

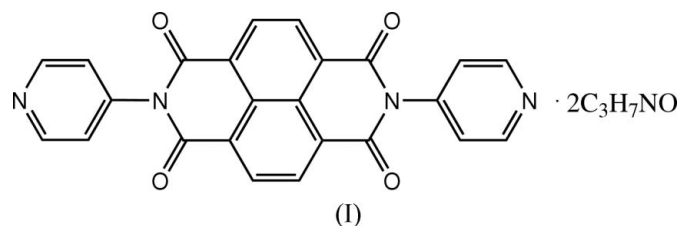
Naphthalene imide compounds are industrially important pigments on the market, side-by-side with perylene imides (Herbst & Hunger, 1993). The title compound, (I), is a *para*-naphthalene imide (*p*-NI) dimethylformamide (DMF) disolvate. We have carried out a series of investigations on H_2 gas sensors utilizing the high proton affinity of organic pigments that have pyridyl rings connected directly to the chromophore (Takahashi & Mizuguchi, 2005). The N atom of the pyridyl ring acts as a strong proton acceptor and plays a decisive role for signal detection. Among these, we have so far determined the structure of *ortho*-dipyridylpyrrolopyrrole (Imoda *et al.*, 2005) and the polymorph of the *para* derivative (Mizuguchi *et al.*, 2002, 2005), as well as *ortho*, *meta* and *para* derivatives of dipyridylperylene imides (Mizuguchi *et al.*, 2005*a,b*; Hino *et al.*, 2005, respectively). The present paper reports the structure of *p*-NI·2DMF, (I).

Fig. 1 shows the *ORTEP* (Burnett & Johnson, 1996) plot of (I). The skeleton of the centrosymmetric *p*-NI is entirely planar, as indicated by a small mean deviation of 0.006 \AA from the least-squares plane of the rings defined by atoms C1–C7/N1. The pyridyl rings are twisted by $69.5(2)^\circ$, in the same direction, out of the plane of the naphthalene imide skeleton. There is a short intermolecular contact of $2.802(8)\text{ \AA}$ between the carbonyl C atom of the skeleton (C4) and the carbonyl O atom of DMF (O3; symmetry code: $1 - x, 1 - y, 1 - z$). This is due to an electrostatic interaction between positively polarized C4 and negatively polarized O3. Because of this, two DMF molecules are bound to the *p*-NI skeleton as shown in Fig. 1.

Fig. 2 shows the molecular arrangement of (I). There are two kinds of stacking columns along the *a* axis, composed of *p*-NI and DMF molecules. Between two DMF columns, the

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polar DMF molecules are arranged in one column in such a way as to cancel the dipole moment of DMFs in the neighboring one, leading to the stabilization of the crystal structure.

Experimental

p-NI was synthesized by reaction of naphthalenetetracarboxylic dianhydride with 4-aminopyridine in dimethylnaphthalene at 473 K for 3 h according to the method in the literature (Herbst & Hunger, 1993). The product was then purified three times by sublimation at 760 K, using a two-zone furnace (Mizuguchi, 1981). Single crystals of (I) were grown from a mixed solvent of acetone and dimethylformamide by slow evaporation.

Crystal data

$C_{24}H_{12}N_4O_4 \cdot 2C_3H_7NO$	$D_x = 1.432 \text{ Mg m}^{-3}$
$M_r = 566.57$	Cu $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 8845 reflections
$a = 5.3890 (8) \text{ \AA}$	$\theta = 3.6\text{--}68.1^\circ$
$b = 12.2012 (17) \text{ \AA}$	$\mu = 0.85 \text{ mm}^{-1}$
$c = 19.998 (3) \text{ \AA}$	$T = 93.1 \text{ K}$
$\beta = 91.764 (10)^\circ$	Needle, gold
$V = 1314.3 (3) \text{ \AA}^3$	$0.50 \times 0.10 \times 0.05 \text{ mm}$
$Z = 2$	

Data collection

Rigaku R-Axis RAPID diffractometer	2241 independent reflections
ω scans	1357 reflections with $F^2 > 2\sigma(F^2)$
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	$R_{\text{int}} = 0.048$
$T_{\text{min}} = 0.625$, $T_{\text{max}} = 0.958$	$\theta_{\text{max}} = 68.2^\circ$
12221 measured reflections	$h = -5 \rightarrow 5$
	$k = -14 \rightarrow 14$
	$l = -23 \rightarrow 23$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.097$	$w = 1/[\sigma^2(F_o^2) + (0.2P)^2]$
$wR(F^2) = 0.309$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.11$	$(\Delta/\sigma)_{\text{max}} < 0.001$
2241 reflections	$\Delta\rho_{\text{max}} = 1.05 \text{ e \AA}^{-3}$
187 parameters	$\Delta\rho_{\text{min}} = -0.73 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

O1—C3	1.246 (5)	N1—C3	1.412 (5)
O2—C4	1.226 (5)	N1—C4	1.434 (5)
C8—N1—C3	114.4 (3)	O1—C3—N1	121.2 (3)
C8—N1—C4	119.0 (3)	O2—C4—N1	120.7 (3)
C3—N1—C4	126.6 (3)		

The H atoms, except for the methyl H atoms, were placed in geometrically idealized positions and constrained to ride on their parent atoms, with $C-H = 0.95 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The methyl H atoms were constrained to an ideal geometry, with $C-H = 0.98 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$, but each group was allowed to rotate freely about its C—C bond. The highest peak of electron density is located 0.94 \AA from H15C. The DMF molecule shows relatively large atomic displacement parameters (Fig. 1), suggesting positional disorder.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2004); program(s) used to solve structure: *SIR2002* (Burla *et al.*,

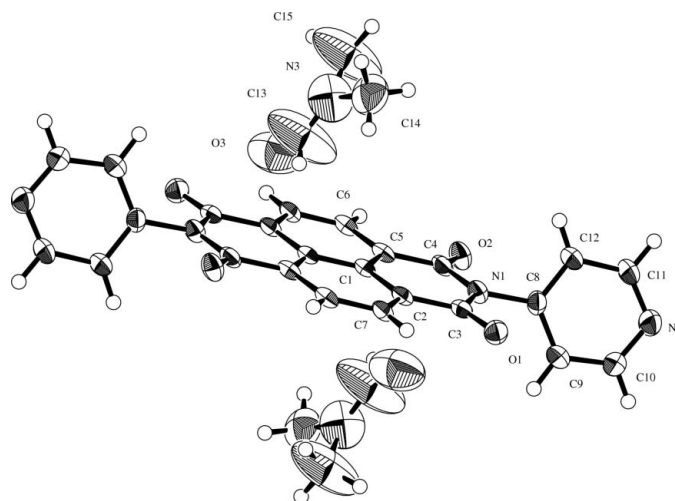


Figure 1

A view of the structure of (I), showing 50% probability displacement ellipsoids. The unlabelled atoms are related to labelled atoms by the symmetry code $(1-x, 1-y, 1-z)$.

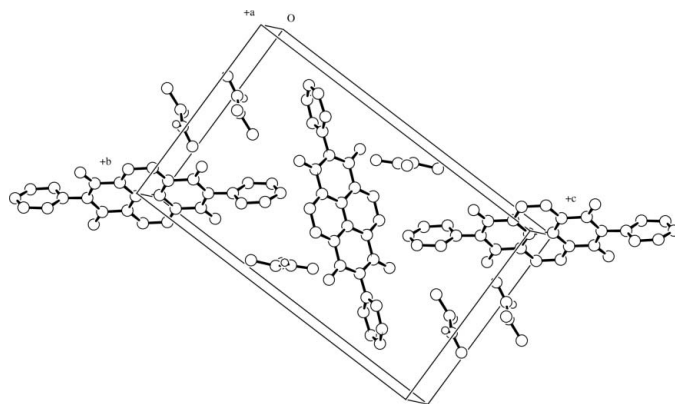


Figure 2

The molecular arrangement of (I).

2003); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP III* (Burnett & Johnson, 1996); software used to prepare material for publication: *CrystalStructure*.

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